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Nine Bottles Experiment By: Jenna, Bailey, and Claire Unknown #4 and #6 The data from the first weeks lab showed that these compounds would form a white precipitate, the data from week two's lab agreed The chemical equation supports this because AgNO_3 and NaCl would combine to
Post Lab 8 The Nine Bottle Problem - 0 em Th c 0 d Ch e ic ...
Chemical Reactions: The Nine Bottle Problem Pre-La ...

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Advice from a Chemistry Tutor: Identifying the Unknown ...

THE EIGHT SOLUTION LAB - science ed The Nine-Bottle Challenge

Solved: You Are Using 9 Different Chemicals For This Labor ...

Advice from a Chemistry Tutor: Identifying the Unknown Solutions Lab. Posted by The Chemistry Lab on 11/27/12 9:26 AM. Tweet; One of the most common general chemistry lab experiments, both in advanced high school classes as well as introductory college courses, is the identification of a series of unknown chemicals. As a chemistry tutor in ...

If the vinegar used in the lab had 5.89 g of acetic acid per 100. mL of vinegar, what volume (in mL) of 0.0998 M NaOH would you use to reach the endpoint of the titration? ... In the reaction of colorless solutions of cobalt(III) Answers · 2. What does it mean if the percent purity of a synthesized aspirin is 100% and the commercial is 90% ...

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Chemical Reactions: The Nine Bottle

Problem Pre-Lab Rb Name Section Number Date: PRE-LAB QUESTIONS e using 9 different chemicals for this laboratory experiment. For each possible 1. You ar reaction between the 9 chemicals, predict the products and write a balanced, complete molecular equation. Look at the products you predicted for the first ...

9.2.1.6 Lab - Using Wireshark to Observe the TCP 3-Way Handshake Answers Lab - Using Wireshark to Observe the TCP 3-Way Handshake (Answers Version) Answers Note: Red font color or gray highlights indicate text that appears in the instructor copy only. Topology Objectives Part 1: Prepare Wireshark to Capture Packets Part 2: Capture, Locate, and [...]Continue reading...

Title Salts and Solubility 3: Solution Equilibrium and Ksp (Inquiry Based) Description I have written a series of five activities using the Soluble Salts simulation to be used throughout the year.

The Nine-Bottle Challenge Name ____ Background: Almost all salts ("salt" is a name for an ionic compound) exist in water solutions as separated ions. For example, a solution of sodium chloride contains Na^+ ions and Cl^- ions, and no NaCl species. When solutions of different salts are mixed, the ionic species present may Anyone do the Nine Bottle Problem? I am given 9 bottles with solutions in each bottle that remain nameless. The object of this is to figure out which bottles contain the following solutions below... Solutions: barium chloride copper (II) sulfate hydrochloric acid lead (II) nitrate magnesium nitrate nickel (II) sulfate sodium carbonate sodium hydroxide zinc chloride Can anyone give me ideas on ...

unknown solutions. You may find that you cannot identify all the solutions and it may be necessary to repeat a combination of solutions to review the results or to try some new combinations. 4. Clean up your lab area and dispose of waste as directed by your instructor LABORATORY REPORT

Chem110 Lab 1 - Experiment 1(The Problem of 9 Unknown ...

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Ok, so I'm sure many of you have heard of this lab. The instructor hands out 9 unknown solutions, and it's your goal to label each. The names have been given to us a day before and we are allowed to research the elements. Copper (II) Nitrate is blue crystals, when hydrated Copper (II) Sulfate is a ...

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THE EIGHT SOLUTION LAB. The reaction of an ion with standard reagents which results in a color change, precipitate or generation of a gas, is a standard method of identifying that particular ion in solution. The experiment given below is an investigation of this principle.

Nine Bottles Experiment by jenna boyle on Prezi

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You are using 9 different chemicals for this laboratory experiment. For each possible reaction between the 9 chemicals, predict the products and write a balanced, complete molecular equation. Look at the products you predicted for the first question. Create a three column table of a) insoluble precipitate formed, b) gas formed, and c) no reaction.

Identification of Unknown Solutions

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Using the 9 chemicals, predict the products, state whether it is a gas, aqueous, or solid, and write a balanced, complete molecular equation. There are a total of 36 equations. 9 Chemicals: NaBr, NaOH, Na₂S, Na₂SO₄, KClO₃, KIO₃, NH₄Cl, BaCl₂, Pb(NO₃)₂ If someone could help me out

with this I'd be completely grateful! I just can't get it :/ I've actually done 29 of them, but can't get the last 7.

Chemical Reactions Help!? | Yahoo Answers

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