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10. How many unpaired electrons are in a sulfur atom? What did you need to answer this question— an orbital diagram or an electron configuration? There are two unpaired electrons. The orbital diagram was necessary (below). 11. Write the electron configuration for zirconium (atomic # = 40). 1s 2s 2p 6s 2 3p 6s 4d 10 4p 6s 4d 12.

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15. Complete the following: Elements ending in d2 are in column number 4 or IVB and those ending in d4 are in column number 6 or VIB. Therefore, elements ending in d3 are in column number 5 or VB. 7 Elements ending in d must be in column number 9 or VIIIB. 16. Notice from question 6 that an element that ends in 3s is in the 3rd row. Also, from questions

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46 ChemQuest 15 Name: _____ Date: _____ Hour: _____ Information: Relating Electron Configurations to the Periodic Table In this section you will see how the periodic table serves as a road map for writing electron configurations. Get your periodic table out and get ready.

46 ChemQuest 15 Name: - Yumpu

the same exponents. Your answer will them have the same exponent as the numbers you add or subtract. Here are some examples. Make sure you understand each step: $4.2 \times 10^6 + 3.1 \times 10^5$ make exponents the same, either a 5 or 6 $42 \times 10^5 + 3.1 \times 10^5 = 45.1 \times 10^5 = 4.51 \times 10^6$

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32 Information: Light Recall that light is a wave. White light is composed of all the colors of light in the rainbow. All light travels at the same speed (c), 3.00×10^8 m/s. (The speed of all light in a vacuum is always equal to

Critical Thinking Questions: Bohr's reasoning

('15 b. Are there any atoms of X that weigh your reported average atomic mass? Explain your answer. NO! 15. The element neon (Ne) is found in three isotopic forms: ^{19}Ne (10% abundant), ^{20}Ne (80% abundant), and ^{21}Ne (10% abundant). Calculate the average atomic mass of Ne. Include proper units with your answer. (

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Therefore, the charge on each X is 1. Here's why: Remember Answer: + that S has a negative 2 charge. There are two "X" ions to balance out the negative 2 charge of S.

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Information: Hund's Rule - S.W.H.S CHEMISTRY

12 9. Formulate a definition for each of the following terms. a) element: Matter that is composed of only one kind of atom. b) compound: Matter that is composed of two or more kinds of atoms chemically combined together

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