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Although the chemical reactions involved in photosynthesis and digestion are different, both chemical reactions are necessary to sustain life. In a chemical reaction, one or more substances (the reactants) change into one or more new substances (the products). Figure 11.1a shows the ingredients for making leavened bread—flour, salt, yeast, and water. Start studying Chapter 11: chemical reactions. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

Chapter 111 Describing Chemical Reactions

a representation of a chemical reaction; the formulas of the reactants (on the left) are connected by an arrow with the formulas of the products (on the right). Write the formulas of the reactants to the left of the yields sign (arrow) and the formulas of the products to the right. first write the skeleton equation.

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11.1 Describing Chemical Reactions 11

Section 11.1 Assessment. Describe the steps in writing a balanced chemical equation. Write the skeleton equation for the following reactions: Heating solid copper(I)-sulfide in the presence of oxygen gas produces pure copper and sulfur dioxide gas. Iron metal and chlorine gas react to form solid iron(III)chloride.

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Every minute of the day, chemical reactions take place—both inside you and around you. •After a meal, a series of chemical reactions take place as your body digests food. •Plants use sunlight to drive the photosynthetic processes needed to produce plant growth.

Chapter 11

Chapter 8.1 : Describing Chemical Reactions 1. Chemical equations and reactions
Chapter 8.1
2. Objectives:
List three observations that suggest that a chemical reaction has taken place.
List three requirements for a correctly written chemical equation.
Write a word equation and a formula equation for a given chemical reaction.
Balance a formula equation by ...

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SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

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11.2 Types of Chemical Reactions > 29 Whether one metal will displace another metal from a compound depends upon the relative reactivities of the

11.2 Types of Chemical Reactions >

Figure 11.1.1: The coffee percolator on the left has been tarnished from exposure to sulfur. Tarnish is the chemical compound silver sulfide. The same percolator on the right has been polished with a tarnish removal product in order to restore its silver finish. (CC BY-NC; CK-12) Another common chemical reaction is the burning of methane gas.

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11.2 Types of Chemical Reactions >

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Every minute of the day, chemical reactions take place—both inside you and around you. •After a meal, a series of chemical reactions take place as your body digests food. •Plants use sunlight to drive the photosynthetic processes needed to produce plant growth.